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1. Name the following compounds. If they begin with hydrogen, name them as acids. Sb(NO₂)₃ antimony (III) nitrite HIO hypoiodous acid (NH₄)₂CO₃ ammonium carbonate LiClO₄ lithium perchlorate HClO₂ chlorous acid HCH₃COO acetic acid Au₃PO₃ gold (I) phosphite Cu₃BO₃ copper (I) borate HNO₂ nitrous acid H₃PO₃ phosphorous acid

Nomenclature #1: Binary Ionic Compounds

WRITING AND NAMING BINARY COMPOUNDS WORKSHEET. WKS 7.1 – Beginning Naming & Formula Writing (2 pages) Name the following ionic compounds. Be sure to check for Roman Numerals! Na₃P. MgBr₂. Ag₂Se. AlCl₃. Ca₃P₂.

WRITING AND NAMING BINARY COMPOUNDS WORKSHEET

Writing and Naming Binary Compounds worksheets are one of the most important parts of a medical test. This worksheet answers key questions that are posed by the medical professional before the patient is taken for the test. This is why it is so important to get the best writing and naming worksheets.

Writing and Naming Binary Compounds Worksheet Answer Key

Type II Binary Ionic Compounds contain Transition metals (including the Group III, IV, V, VI metals, except for Al) with non-metal ions. Show the correct name for the following compounds. Give correct formulas for these Type II binary compounds

Type II Binary Compound Answers - MiraCosta College

A B; lithium fluoride: LiF: lithium chloride: LiCl: lithium bromide: LiBr: lithium iodide: LiI: lithium oxide: Li₂O: lithium sulfide: Li₂S: lithium nitride: Li₃N ...

Quia - Binary Ionic Compounds

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Naming Compounds Tutorial and Worksheet Since we use different methods in naming binary covalent (molecular) compounds and ionic compounds, the first step

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in naming or writing the formula of a compound is to determine which of the 2 compound classes it belongs. Name the first element: boron 2. . Worksheet 4.

Naming binary compounds worksheet answers

Binary ionic compounds always have 2 types of atoms. 3. When naming an ionic compound, you should name the metal first and the nonmetal second (metal or nonmetal). 4. Circle one: when naming a binary ionic compound, the ending of the first element does/does not change. 5. When naming a binary ionic compound the nonmetal always has an -ide ending.

Naming compounds Inquiry (1).docx - Naming compounds ...

Binary ionic compounds worksheet #2 Name Carlos Perez P____ Date____ A. Name the following compounds: 1. LiCl Lithium Chloride 2. Cu O Copper (II) oxide 3. FeBr 2 Iron (II) Bromide 4. KH Potassium Hydride 5. MgI 2 Magnesium Iodide 6. K 2 O potassium Oxide 7. ZnO Zinc oxide 8. FeF 3 Iron (III) Flouride 9. CaS Calcium Sulfide 10. Mn 2 S 3 ...

Binary ionic compounds worksheet 2.docx - Binary ionic ...

If a binary compound is composed of two nonmetals, it is a covalent/molecular compound. Use the appropriate prefixes to name these simple compounds. Mono- is only used as a prefix on the second element in the formula. The ending for the second element is always removed and replaced with -ide. 1 = mono- 2 = di- 3 = tri- 4 = tetra- 5 = penta- 6 = hexa- 7 = hepta- 8 = octa- 9 = nona- 10 = deca-.

Naming Binary Covalent Compounds Flashcards | Quizlet

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Formulas and Nomenclature Binary Ionic Compounds Worksheet ...

Give correct formulas for these Type I binary compounds; calcium iodide: magnesium phosphide: calcium hydride: sodium chloride: magnesium fluoride: barium oxide: cadmium bromide: aluminum arsenide: sodium nitride: calcium sulfide: rubidium oxide: potassium selenide: barium nitride: sodium iodide: lithium chloride: lithium sulfide: silver sulfide: calcium carbide: aluminum nitride

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