

Handbook Of Refractory Carbides Nitrides Properties Characteristics Processing And Apps Materials Science And Process Technology

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Refractory carbides and nitrides are materials with a high melting point, arbitrarily selected as greater than 1800°C and must have a high degree of chemical stability. Carbides and nitrides can be divided into several types depending on their different physicochemical structures and characteristics.

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Handbook of Refractory Carbides and Nitrides: Properties, Characteristics, Processing and Applications (Materials Science and Process Technology) 1st Edition by Hugh O. Pierson (Author)

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Handbook of Refractory Carbides and Nitrides Figure 3.6: Cloud representation of the sp^3 hybrid orbital. The energy required to accomplish the sp^3 hybridization and raise the carbon atom from the ground state to the corresponding valence state V, is 230 kJ mol^{-1} .

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title = {Handbook of refractory carbides and nitrides: Properties, characteristics, processing and applications}, author = {Pierson, H O}, abstractNote = {This reference work provides a complete review of the structure properties, processing and applications of refractory carbides

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and nitrides. The contents include: the refractory carbides; interstitial carbides, structure and composites; titanium, zirconium, and hafnium carbides; vanadium, niobium and tantalum carbides; chromium, molybdenum

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The refractory carbides and nitrides are reviewed together in one book because : • They form two families of closely related materials which

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have similar atomic structures and chemistry. Carbon and nitrogen are next to each other the second period of the Table of the Elements and, in many cases, the carbides and nitrides form solid solutions (known as carbonitrides)

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In cemented carbides, typically 60–95 vol.% of the carbide phase is bonded by a more ductile metal, invariably a cobalt, nickel or iron alloy. The most important refractory carbides are WC, TiC and TaC; annual world production of these carbides is about 20 kt, 1.5 kt and 0.35 kt, respectively.

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