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1. The alkali metals all have a single s electron in their outermost shell. In contrast, the alkaline earth metals have a completed s subshell in their outermost shell. Answer Key Chapter 18 - Chemistry 2e | OpenStax

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13. Possible answers include: cylinder filled with. clear liquid, 360 mL, density = 1 g/cm³, mp = 0 ° C, bp = 100 ° C. 14. Possible answers include: flesh color (student. specific), solid filled with liquids, 7 cm, moves. through muscle and ligament connections. 15. Possible answers include: steel, easily magnetized, 3 1/4 cm × 3/4 cm, silver, solid

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Study Guide and Reinforcement - Answer Key. 18 Answer Key
ANSWER KEY Chapter 18 Section 2 Section 1 Answers 1-3 are

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interchangeable. 1. repulsive 2. greater 3. long 4. strong 5. short 6. neighbors 7. protons 8 12. number of protons 5. 8 g 13. number of electrons 7. 2 g 14. chemical properties 8. 1 g Answers 15-16 are interchangeable.

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Section 18.2 Assessment page 649 17. Describe the contents of dilute aqueous solutions of the strong acid HI and the weak acid HCOOH.

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The solution of HI contains only H_2O and I^- and water molecules. The solution of HCOOH contains H^+ and HCOO^- ions, and HCOOH and H_2O molecules. 18. Relate the strength of a weak acid to the strength of its conjugate base.

Acids and Bases Acids and Bases - Weebly

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Guide Answers chapter 1 chemistry matter change Flashcards and ... A bond characterized by the sharing of one or more pairs of electrons between two atoms. A force holding metal atoms together and characterized by free or delocalized electrons.

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