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REVIEW. Stoichiometry.

SECTION 1. SHORT ANSWER

Answer the following

questions in the space

provided. 1. \_\_\_\_\_ The

coefficients in a chemical  
equation represent the.

(a) masses in grams of all  
reactants and products.

(b) relative number of moles  
of reactants and products.

(c) number of atoms of each  
element in each compound in

a ...

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#### Chapter 9 Stoichiometry

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Modern Chemistry 77

Stoichiometry CHAPTER 9

REVIEW Stoichiometry SECTION

3 PROBLEMS Write the answer  
on the line to the left.

Show all your work in the  
space provided. 1. \_\_\_\_\_ The  
actual yield of a reaction  
is 22 g and the theoretical  
yield is 25 g. Calculate the  
percentage yield. 2. 6.0 mol  
of  $N_2$  are mixed with 12.0  
mol of H

#### CHAPTER 9 REVIEW

#### Stoichiometry

CHAPTER 9 REVIEW

Stoichiometry SECTION 3

PROBLEMS Write the answer on

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## Review Stoichiometry

the line to the left. Show

all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of  $H_2$  according to the following equation:  $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$

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stoichiometry (which you studied in Chapter 3) deals with the mass relationships of elements in compounds. Reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

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#### **Answer Key**

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Reactions SECTION 82 SHORT  
ANSWER Answer the provided a  
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Stoichiometry SECTION 9-3  
PROBLEMS Write the answer on

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involving the mass  
relationships of elements in  
compounds. Reaction  
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involving the mass  
relationships between  
reactants and products in a  
chemical reaction. Mole  
Ratio ...

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Percent Yield 2 Definitions.  
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relationships of elements in  
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VOCABULARY\*\* Composition  
stoichiometry. Reaction  
stoichiometry. Mole ratio.  
Limiting reactant.  
calculations involving the  
mass relationships of  
elements in c... calculations  
involving the mass

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relationships between  
reactant... a conversion  
factor that relates the  
amounts in moles of any t...  
the substance that controls  
the quantity of product that  
...

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Modern Chemistry Chapter 9  
Review Stoichiometry Section  
2... Reaction stoichiometry  
uses molar relationships to  
determine the amounts of  
unknown reactants or  
products from the amounts of  
known reactants or products.

## **Chapter 9 Review Stoichiometry Section 2**

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### Answers

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